

## The Charge-transfer Band in the Ultraviolet Spectrum of Cyclopropyl Ketones

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*Summary* The charge-transfer band, predicted to occur in the ultraviolet spectrum of cyclopropyl ketones, is observed at about 180 nm, and has a molar extinction of *ca.* 4000.

AN "Extended-Hückel" analysis<sup>1</sup> indicates that the energy of the degenerate pair of highest-occupied molecular orbitals in cyclopropane ( $\Delta$ ) is comparable with that of the carbonyl  $\pi$  and lone-pair ( $n$ ) levels. This suggests that the u.v. spectrum of cyclopropyl ketones would contain, somewhere

between the  $n \rightarrow \pi^*$  (*ca.* 300 nm) and  $\pi \rightarrow \pi^*$  (*ca.* 150 nm) absorptions,<sup>2</sup> an additional band,  $\Delta \rightarrow \pi^*$ , representing charge-transfer from the three-membered ring, to C=O.

To locate the novel transition, we carried out MO calculations of the type used in the study of cyclopropane<sup>3</sup> and phenyl-ketones.<sup>4</sup>  $\Delta \rightarrow \pi^*$  is predicted to lie at 176–178 nm, with oscillator strength in the range 0.00–0.08, dependent upon conformation.

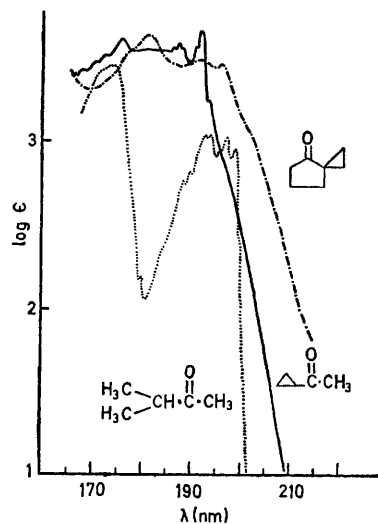
We have observed† this transition in the vapour-spectra of several cyclopropyl ketones (*e.g.*, cyclopropylmethyl

† Spectra were recorded on a McPherson vac.-u.v. spectrometer (model 225), a discharge lamp with H<sub>2</sub> continuum serving as source.

ketone and spiro[2,4]heptan-4-one,<sup>‡</sup> see Figure). As expected, it is situated at *ca.* 180 nm, in the gap between two bands that characterize all ketones<sup>5</sup> (the band just below 200 nm is the one discerned, in solution spectra, at *ca.* 200 nm).<sup>6</sup> It resembles in intensity [ $\epsilon$  (mol) *ca.* 4000] the bands that flank it, forming with them a pattern that is quite distinct from that of non-cyclopropyl ketones (*e.g.*, that of 3-methylbutan-2-one, see Figure).

A more detailed analysis of spectra and MO results will be given elsewhere.

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<sup>‡</sup> A sample of spiro[2,4]heptan-4-one was kindly supplied by Prof. J.-M. Conia, and used as such. 3-Methylbutan-2-one and cyclopropylmethyl ketone were purified by g.l.c., with a 20 ft × 0.75 in column of 10% carbowax 20 M on acid-washed Chromosorb W (60/80 mesh).

<sup>1</sup> R. Hoffmann, *Tetrahedron Letters*, 1965, 3819.

<sup>2</sup> G. Berthier and J. Serre, in 'The Chemistry of the Carbonyl Group', ed. S. Patai, Interscience, London, 1966, p. 1.

<sup>3</sup> A. Y. Meyer, *Theor. Chim. Acta*, 1971, **22**, 271.

<sup>4</sup> A. Y. Meyer, *Theor. Chim. Acta*, 1968, **9**, 401.

<sup>5</sup> E. E. Barnes and W. T. Simpson, *J. Chem. Phys.*, 1963, **39**, 670.

<sup>6</sup> E. M. Kosower, *J. Amer. Chem. Soc.*, 1958, **80**, 3261.